Chemical and microbiological processing for modification of 2:1 sheet silicate structure and properties

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Developing advanced ceramics materials, used in industrial, building and domestic products, will continue to be important and closely connected with reducing our energy consumption and our impact on the environment.

A new ceramics product can be characterised with:

- Reduced sintering temperature
- Reduced thermal conductivity
- Optimal provision of ceramic properties
- Optimal provision of mechanical properties

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The aim of the work

The aim of this work is to show impact of NaOH of different concentrations and biological impact (Pseudomonas fluorescens bacterium treatment) on illite clay in an attempt to change or “ruin” specific network-forming linkages among Si-O and Al-O units.

Paliek!
To investigate and to analyse ceramic and mechanical properties.

To perform cylindrical clay samples forming and sintering at temperatures 100, 300, 600 and 700°C.

To determinate the crystalline phase changes both for chemically and biologically treated clay mixes and sintered ceramic samples.

To activate with 1, 3, 4 and 6 M NaOH solutions and with Pseudomonas fluorescens bacterium AM-PS11.

To prepare the clay powder by granulation and screening.

To investigate and to analyse ceramic and mechanical properties.
Experimental methods

Yield of an average clay sample from 2-3m deep layer of clay

To grind and screen clay sample

Preparation of powdery samples for treatment with 1M, 3M, 4M and 6M NaOH solutions

Successive treatment of samples with NaOH solutions for 24h

Study of the treated clay: XRD, IR-spectra, DTA

Preparation and sintering of the samples in temperature range from 100°C to 700°C

Investigation of ceramic properties (total porosity, bulk density, shrinkage) as well as compressive strength

Paliek!
Experimental methods

Yield of an average clay sample from 2-3m deep layer of clay → To grind and screen clay sample → Sterilization of clay 121°C, 15min; 24h cultivation of Ps. Fluorescens

Incubation of clay suspension with culture liquid

Study of the treated clay: XRD, IR-spectra, DTA

Preparation and sintering of the samples in temperature range from 100°C to 1000°C

Investigation of ceramic properties (total porosity, bulk density, shrinkage) and flowability
Clay deposits are widely spread and most important mineral wealth in Latvia, they can be found at various stratum depths from Cambrian to Quaternary geological periods. The most widely spread clays are illites - they form 70-80% of the total volume with slight admixture of chlorites, smectites and kaolinites.

The deposits of geological periods that contains clayey rock:

- Cambrian
- Ordovican
- Quaternary
- Sillurian
- Devonian
- Jurassic
- Triassic

Most practical importance is for the clay from Devonian and Quaternary periods.
The creation of Quaternary sediments began 70 thousand years ago. Clayey rocks are the youngest and they cover all the territory of Latvia.

Fig. 1. The thickness of Quaternary sediments according to the data of geological mapping.
Structure of clay

- Clay is the sedimentary rocks, that start forming from magmatic and metamorphic rocks decomposition products.
- Main groups of clays are clay’s minerals - kaolinite, illite, montmorillonite.
- In structural they are layered aluminosilicates, those main structure element are repeatable tetrahedral Si-O and octahedral Al-O sheets.

a) Diagrammatic representation of different clay minerals
b) Cristal lattice of illite
Processes occurring during sintering

Traditionally clay is used for production of domestic, industrial, building products and art objects. Firing them at maximum temperatures, between the intervals of 950 -1050 °C, is a part of production process.

The main reactions occurring by sintering

the dominating reactions taking place in clays in the interval of temperatures above 600-700 °C are as follow:

- separation structural water from illite structure:
  
  \((K, H_3O)Al_2(OH)_2[(Si, Al)_4O_{10}]n\ H_2O \rightarrow K[AlSi_3O_8]+ Al_2O_3 +SiO_2 +H_2O\)

- simultaneously at 620 – 912 °C carbonates of the clay decompose:
  
  \(\text{CaCO}_3 \cdot \text{MgCO}_3 \rightarrow \text{CaO} + \text{MgO} + \text{CO}_2 \uparrow\)

- \(\text{Fe}^{3+}\) reduces to \(\text{Fe}^{2+}\) ; \(\text{Fe}_3\text{O}_4 \rightarrow \text{FeO} +\text{O}_2 \uparrow\)

- forming a eutectic mixture/liquid phase – the material shrinks becomes denser and harder
Processes occurring in clay under alkaline activation

Atstāt!

The chemical activation of clay minerals/clays with alkaline hydroxides can form materials with fundamental different properties, such as a ceramic material with sufficient strength, at temperature which is less than the traditional firing temperature.

Schematic representation of the structural changes of the clay minerals under the impact of alkaline can be described with reactions:

- \((K,H_3O)Al_2(OH)_2[AlSi_3O_{10}].nH_2O + 3K^+ +3OH^- (H_2O)\)
  illite

- \(3 Al(OH)_3 +3Si(OH)_4 + 4K^+ \) or \(K_4Al_3Si_3O_{12} + n H_2O\)
  zeolite similar to ceolite

- dissolution of kaolinite and forming soluble Si-OH and Al-OH groups
  a) \(Al_2Si_2O_5(OH)_4+6OH^- +H_2O \rightarrow 2Al(OH)_3^- +2H_2SiO_4^{2-}\)
  kaolinite
  b) formation of feldspathoid - hidrosodalite:
  \(6Al(OH)_3^- +6H_2SiO_4^{2-} + 6Na^+ \rightarrow Na_6Si_6Al_6O_{24} + 12 OH^- + 12H_2O\)
  zeolite similar to ceolite
Processes occurring in clay under biological treatment

The biological treatment of clay minerals/clays can impact on rheology of clays.

The hypothesis of processes in clay by biological treatment:
- flowability-degradation illite clay by change water content;
- anionic and cationic exchange capacities;
- technological characteristics such as shrinkage, plasticity and sensitivity to drying;
- the chemical and physical properties and phase changes (less possible)

Noņemt!
Experimental procedure

The clay from the Laza quarry (Kurzeme district, Latvia) was taken from 2-3m depth layer.

Different concentration of NaOH solutions: 1M, 3M, 4M and 6M.
### Chemical composition of the clay (wt.%)

<table>
<thead>
<tr>
<th></th>
<th>SiO₂</th>
<th>Al₂O₃</th>
<th>CaO</th>
<th>MgO</th>
<th>Fe₂O₃</th>
<th>TiO₂</th>
<th>Na₂O</th>
<th>K₂O</th>
<th>I.O.I.*</th>
</tr>
</thead>
<tbody>
<tr>
<td>Values</td>
<td>46.87</td>
<td>18.74</td>
<td>6.62</td>
<td>3.25</td>
<td>7.06</td>
<td>0.59</td>
<td>0.41</td>
<td>3.62</td>
<td>12.84</td>
</tr>
</tbody>
</table>

*Loss on ignition

### Grading composition of the clay (wt.%)

<table>
<thead>
<tr>
<th>Fraction</th>
<th>Chemical Composition</th>
</tr>
</thead>
<tbody>
<tr>
<td>Sand fraction &gt; 0.05 mm</td>
<td>1.33</td>
</tr>
<tr>
<td>Aleurit (dust) fraction 0.05 - 0.005 mm</td>
<td>8.67</td>
</tr>
<tr>
<td>Clay mineral fraction</td>
<td></td>
</tr>
<tr>
<td>&lt;0.005 mm</td>
<td>90.00</td>
</tr>
<tr>
<td>int.all &lt; 0.002 mm</td>
<td>63.60</td>
</tr>
</tbody>
</table>
- *Pseudomonas fluorescens* are Gram-negative rod shaped bacteria that inhabit soil, plants, and water surfaces;
- These bacteria are easy to culture, as they grow on minimal media and at 25-30°C and grow fairly quickly;
- Samples of *Pseudomonas fluorescens* for study are available from some companies that supply stabilized bacteria for research in samples designed to be free of contaminants.
- As these bacteria produce a lot of around production, they is very economically advantageous.
**XRD**

It must be noted that changes of crystalline phases intensity for raw clay $L_{fr.}$ by the microbiological treatment is not observed.
Fig. 2. IR-spectra of not treated clay L and treated with NaOH solutions
Fig. 3. IR-spectra of not treated clay L and treated by Ps. Fluorescens bacteria
Fig. 4. DTA curves for not treated clay L and treated with NaOH solutions.
Fig. 5. DTA- curves for not treated clay L and treated by Ps. Fluorescens bacteria
Results and discussion

Fig. 6. Total shrinkage of ceramics samples in dependence of molar concentration of NaOH solutions.
Fig. 7. Total porosity of ceramics samples in dependence of molar concentration of NaOH solutions.
Fig. 11. Differential porosity development of sintered ceramic samples L on molar concentration of NaOH solution.
Results and discussion

Fig. 8. Bulk density of ceramics samples in dependence of molar concentration of NaOH solutions.
Fig. 9. Compressive strength of ceramics samples in dependence of molar concentration of NaOH solutions.
Fig. 10. Comparative characterization of flowability-degradation of three illite clay samples in dependence on bacteria *Ps. Fluorescens AM PS11* concentrations.
Results and discussion

Fig. 11. a) and b). Bulk density and shrinkage change on sintering temperature.
- Alkaline with a sodium hydroxide of not dehydroxylated layered aluminosilicate – illite clay has been studied.

- This study shows that alkaline activation of raw illite clay by sodium hydroxide of different concentration at room temperature leads to the slight structural changes mainly characterized by water link changes in illite structure together with decrease of diffraction peaks of all crystalline phases for illite and kaolinite and with illite structural water losses characterized by DTA.

- It is shown on considerable influence of chemically treatment on some properties of sintered samples in temperature range from 100 °C to 700 °C ceramic samples. In particular there are noticeable changes of total porosity and bulk density as well as increase of compressive strength in the whole sintering temperature interval by reaching maximum values of > 30MPa at 700°C.
Thank You